

(52)

$$Q = Q_{\text{ice}} + Q_{\text{ice} \rightarrow \text{water}} + Q_{\text{water}}$$

$$Q = c_{\text{ice}} m \Delta T + m L_f + c_w m \Delta T$$

$$4.11 \times 10^6 \text{ J} = (2.00 \times 10^3 \frac{\text{J}}{\text{kg}^\circ\text{C}}) 10.0 \text{ kg} [0 - (-10^\circ\text{C})] \\ + (10.00 \frac{\text{kg}}{\text{kg}}) (33.5 \times 10^4 \text{ J/kg}) + 4186 \frac{\text{J}}{\text{kg}^\circ\text{C}} 10 \text{ kg} (T - 0)$$

$$4.11 \times 10^6 = 2.00 \times 10^5 + 33.5 \times 10^5 + 41860 T$$

$$41860 T = 5.6 \times 10^5 \Rightarrow T = 13^\circ\text{C}$$

(50)

$$Q = c m \Delta T \Rightarrow \Delta T = \frac{Q}{c m} = \frac{Q}{c \rho V} = \frac{Q}{c \rho A L}$$

Contracts according to: $\Delta L = \alpha L \Delta T$

$$\Delta L = \alpha L \frac{Q}{c \rho A L} = \frac{\alpha Q}{c \rho A}$$

Stress is given by $F = Y A \frac{\Delta L}{L}$

$$= Y A \frac{\alpha Q}{L c \rho A} = \frac{\alpha Q Y}{L c \rho}$$

$$F = \frac{(12 \times 10^{-6} \text{ }^\circ\text{C}^{-1})(3300 \text{ J})(2.0 \times 10^{11} \frac{\text{N}}{\text{m}^2})}{(452 \frac{\text{J}}{\text{kg}^\circ\text{C}})(2.0 \text{ m})(7860 \text{ kg/m}^3)} = 1.1 \times 10^3 \text{ N}$$

(73)

$$\% \text{ RH} = \frac{\text{partial pressure}}{\text{Saturation Vapor pressure}} \times 100$$

$$= \frac{\text{Dew point Vapor pressure}}{S. V. P} \times 100$$

$$= \frac{1400 \text{ Pa}}{4200 \text{ Pa}} \times 100 = 33\%$$